

#### **Dash Weeks**

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MEASUREMENT

LABORATORY

National Institute of Standards and Technology U.S. Department of Commerce

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# EXPANDING THE CO<sub>2</sub> INFRASTRUCTURE

- The state of the CO<sub>2</sub> equation of state
- Effects of impurities on thermodynamics
- Erosion/corrosion implications
- In-situ mechanical material properties
- NIST involvement overview

### EXPANDING THE CO<sub>2</sub> INFRASTRUCTURE

f(p,V,T)=0

Ideal Gas Law pV = RT1834

Non-Ideal Gases (multiple cubic equations)  $p = \frac{RT}{V_m - b} - \frac{a}{V_m^2} \quad \text{van der Waals 1873}$   $p = \frac{RT}{V_m - b} - \frac{a\alpha}{V_m(V_m + b)} \quad \text{Soave-Redlich-Kwong 1972}$   $p = \frac{RT}{V_m - b} - \frac{a\alpha}{V_m^2 + 2bV_m - b^2} \quad \text{Peng-Robinson 1976}$ 

<u>There are many more versions and modifications but the most recent EOS are adequate for many</u> <u>engineering purposes.</u>

The state of the CO<sub>2</sub> Equation of State

# STATE OF THE CO<sub>2</sub> EOS

#### Need:

- Accurate Pressure-Volume-Temperature, enthalpy behavior, and phase equilibria is essential for design.
- Better accuracy than simple cubic equations especially near phase boundaries and critical points.
- Viscosity and thermal conductivity correlations, which require calculating density as a function of temperature and pressure.

**Solution:** Current Reference EOS: Span and Wagner (1996)

- Experimentally educated
- Multi-parameter
- Substance-specific and at liquid-like densities

Incorporated in NIST's REFPROP

The Span and Wagner EOS has 42 terms of different types. It is computationally complex and too slow for some applications. There are numerical problems in mixture models and near critical points.

# NIST CO<sub>2</sub> EOS EVOLUTION

- New experimental data is available since 1996; the speed of sound is more accurate than heat capacity which reduces uncertainty.
- We have a better understanding of theoretical constraints, and the model is a better representation of super-critical and high-temperature behavior.
- Can now perform state-of-the-art molecular modelling and calculations for heat capacity, thermal conductivity, and viscosity.
- The latest EOS optimization techniques retain high accuracy with fewer terms and improved computational efficiency (factor 1.5 – 2)

RUHR UNIVERSITÄT BOCHUM

This work is part of an international collaboration with Ruhr University in Bochum Germany. The collaborators are:

- E.W. Lemmon, A.H. Harvey, and I.H. Bell, Applied Chemicals and Materials Division, NIST
- T. Eckermann, R. Span, and M. Thol, Ruhr University Bochum.

### REFPROP

# **REFerence Fluid** Thermodynamic and Transport PROPerties (NIST23)

Define Mixture Avalge halo T:3-Butadene (Buta-1,3-dene) T:3-Butadene (Buta-1,3-dene) T:3-Butadene (Buta-1,3-dene) T:3-Butadene (Buta-1,3-dene) T:2-Dimethylottane 2-2-Dimethylottane 2-2-Dimethylottane 2-3-Dimethylottane 3-Mathylopertane Acchore (Propanone) Acchore (Elsne) Actione (Butane) Cathon missource Cathon missource Cathon missource Cathon missource Cathon missource Colorie Chlorolene (Sa-2-Butane) Colorene Colorene Colorene Colorene Di (Dodcamethyloctolertasioxane) D5 (Docamethyloctolertasioxane) D6 (Dodcamethyloctolertasioxane) D6 (Dodcamethyloctolertasioxane) D6 (Dodcamethyloctolertasioxane) D6 (Dodcamethyloctolertasioxane) Decane Dictolecethane (12-Dinlocethane) Dictolecethane (12-Dinlocethane) Dictolecethane (12-Dinlocethane) Dictolecethane (12-Dinlocethane) Dictolecethane (Dimethyloctolertasioxane) Decane Dictolecethane (12-Dinlocethane) Dictolecethane (12-Dinlocethane) Dictolecethane (Dimethyloctolecatioxane) Dictolecethane (Dimethyloctolecatioxane) Dictolecethane (Dimethylocethane) Dimethyl (artichylocethane) Dimethyl (artichylocethane) Dimethy		≜dd →       <-Bernove	Selected mixtu Butane (n- Carbon diox	re components Butane) ide pecify Mixture Com Name (Butane/Carb	xepponents ane)  ify Masture Composition ge [Butane/Carbon dioxide Composition		
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#### Simple Workflow and GUI

- Select Fluid/Mixture
- Select Units & Properties
- Select calculation(s)
- Plot/Export



 REFPROP is a computer program that can provide the thermophysical properties of common industrial fluids and their mixtures, <u>including CO<sub>2</sub> and</u> <u>CO<sub>2</sub> mixtures</u>.

- Computer program sold by NIST Standard Reference Data <u>https://www.nist.gov/srd/refprop</u>
- It may be used on a Windows machine with the graphical user interface (GUI) or one can interface the program with 3<sup>rd</sup> party software and programs such as Python, Excel, Matlab, Mathematica, etc. using the supplied DLL.

"The NIST REFPROP Database for Highly Accurate Properties of Industrially Important Fluids", Marcia L. Huber, Eric W. Lemmon, Ian H. Bell, and Mark O. McLinden, Ind. Eng. Chem. Res. 2022, 61, 42, 15449-15472 <u>https://doi.org/10.1021/acs.iecr.2c01427</u>

#### **MATERIAL MEASUREMENT LABORATORY** Applied Chemicals and Materials Division – Thermophysical Properties

### CO<sub>2</sub> PHASE DIAGRAM



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### CO<sub>2</sub> PHASE DIAGRAM UNCERTAINTY



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# MIXTURE OR CONTAMINANT (?)

Small concentrations result in small changes in thermodynamic properties

- 1. What is a significant concentration (change)?
- 2. What are the limits of detectability?
- 3. How accurately can we measure concentration?

### CO<sub>2</sub> Measurements and Reference Materials

https://www.nist.gov/programs-projects/carbon-dioxide-measurements-and-reference-materials



<u>Contaminants also impact:</u> Mechanical Material Properties Corrosion/Erosion Health & Safety Operations

What are acceptable levels of contaminants to maintain pipeline integrity and public safety?

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# **RANGE OF CONDITIONS IN THE LITERATURE**



**Duration** hours  $\rightarrow 1$  year





**Temperature** 12 °C – 265 °C

> **Contaminants**  $H_2O, H_2S, O_2, NO_2, SO_2$ (often in combination)

**Result**: Reported corrosion rates vary over four orders of magnitude (0.01 – 100 mm/year).

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### **IMPORTANCE OF FLOW RATE**



L. Wei, X. Panga, K. Gaoa, Effect of flow rate on localized corrosion of X70 steel in supercritical CO2 environments, Corr. Sci. 136 (2018)

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# NIST CO<sub>2</sub> TEST FACILITY







- State-of-the-art test facility for SCCO<sub>2</sub> environmental testing within a walk-in fume hood for H<sub>2</sub>S testing capability with remote monitoring and continuous data-logging.
- Static and dynamic test applications to evaluate flow-induced damage on multiple alloys simultaneously.
- Multiple autoclaves up to 206 bar and 500 °C
- IGS Analyzer (FTIR) for constituent monitoring

Real conditions include turbulent mixing and contaminants are replenished (locally). Lab conditions have little to no mixing and a limited contaminant supply (locally).

# NIST CO<sub>2</sub> CORROSION TESTS



NACE TM0169 ASTM G31

# NIST CO<sub>2</sub> FLOW Simulations





Test coupons are 75 x 19 mm and 3 mm thick

#### ASTM G184

Concept modified for use in an autoclave with a magnetic drive.



**Flow Rate** 

Higher impurity levels are expected to produce thicker corrosion products which could lead to spallation and erosion-corrosion.

# NIST CO<sub>2</sub> IN-SITU MECHANICAL TESTS





ASTM E1681





NACE TM0177 Method B



### Advancing Carbon Dioxide Removal

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#### Examples of focus areas include:

- Direct Air Capture Sorbent Materials Characterization
- Low Carbon Cements and Concretes Consortium
- Next Generation Seawater CO<sub>2</sub> Reference Materials
- <u>CO<sub>2</sub> Measurements and Reference Materials</u>
- Ocean Color Satellite Calibration
- Standards to Underpin MRV
- Standards for Carbon Markets and Accounting
- CO<sub>2</sub> Pipeline Material Reliability
- Convergence with
  - GHG Measurements
  - Circular Economy
- NIST collaborates with industry, academia, standards, and government organizations on a broad array of **research, measurements, and standards** to advance CDR, carbon accounting, and decarbonization.

MATERIAL MEASUREMENT LABORATORY







CO2 adsorption via in-situ STEM-EELS imaging





US Patent US9431211B2 Hybrid electron microscope; Ultramicroscopy 150, 10 (2015), Nature Materials 18 (6), 614–619 (2019) Microscopy and Microanalysis 27 (S1), 800-801 (2021)

#### **Carbon Accounting and Decarbonization Program**

NIST's Mission is to promote U.S. innovation and industrial competitiveness by advancing measurement science, standards, and technology in ways that enhance economic security and improve our quality of life.



Develop critical measurement science to accelerate innovation, scalability, & reduce uncertainties

#### Benchmark Measurements Data & Materials



Facilitate **rigor and reproducibility** across measurement ecosystems

#### **Documentary Standards**



Support industry and Federal use of voluntary consensus standards

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