



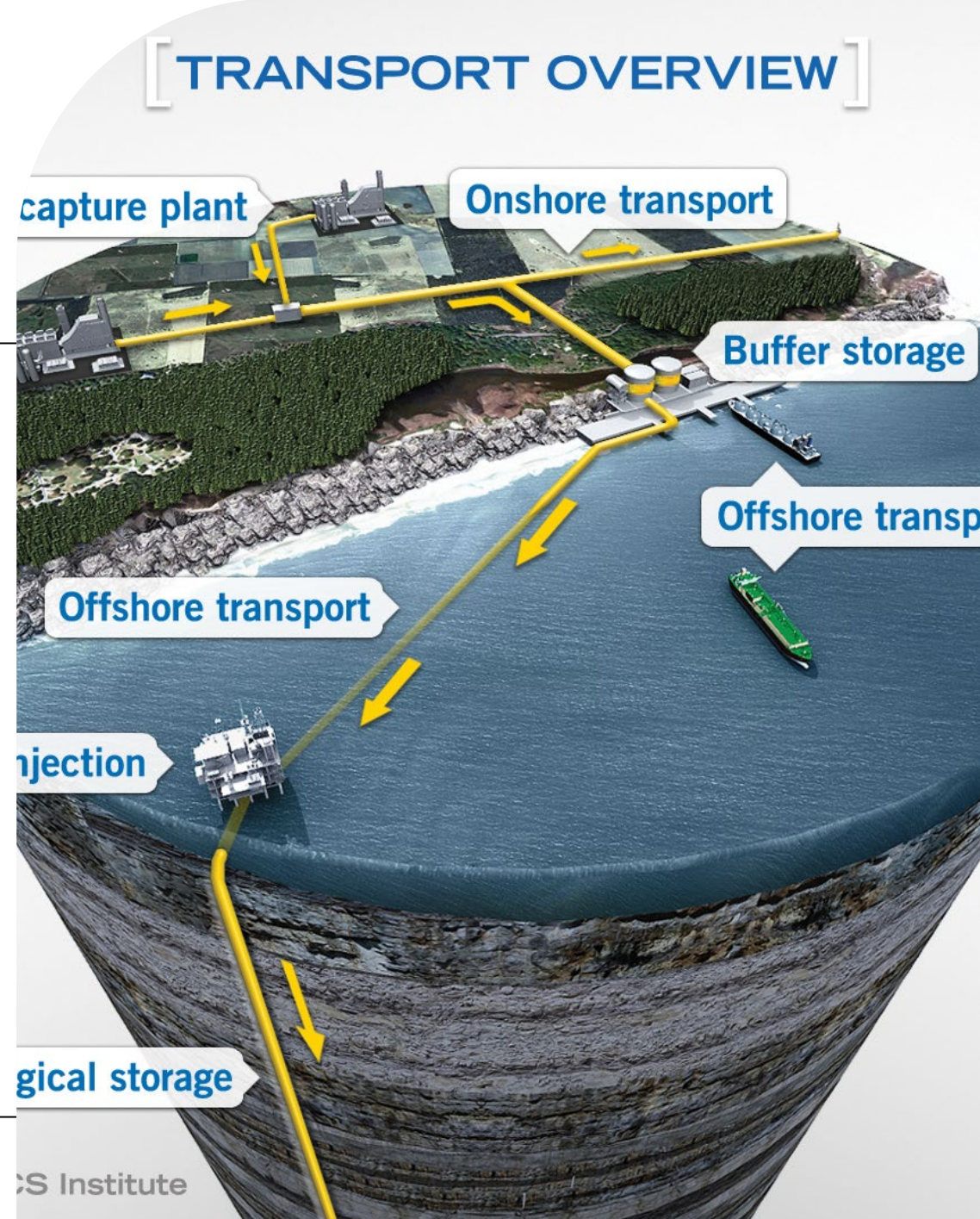
# Odorization technology

For improved CO<sub>2</sub> onshore transport safety



*Pipeline Safety Research and Development Forum 2023*

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Gas odorant Technical Manager

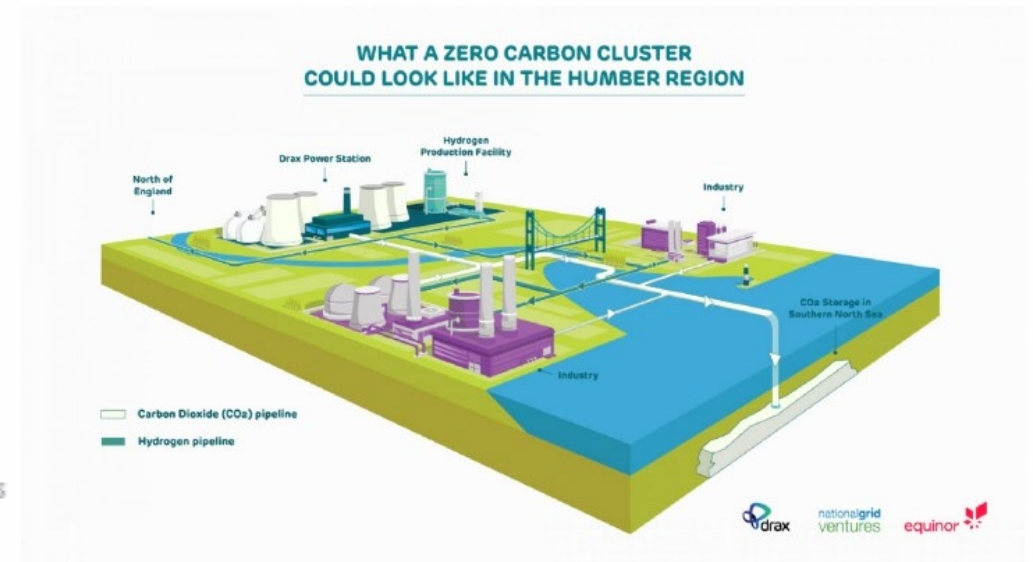


# CARBON NEUTRALITY OBJECTIVES AND CARBON CAPTURE

→ IPCC scenario 1,5° : toward Carbon Neutrality

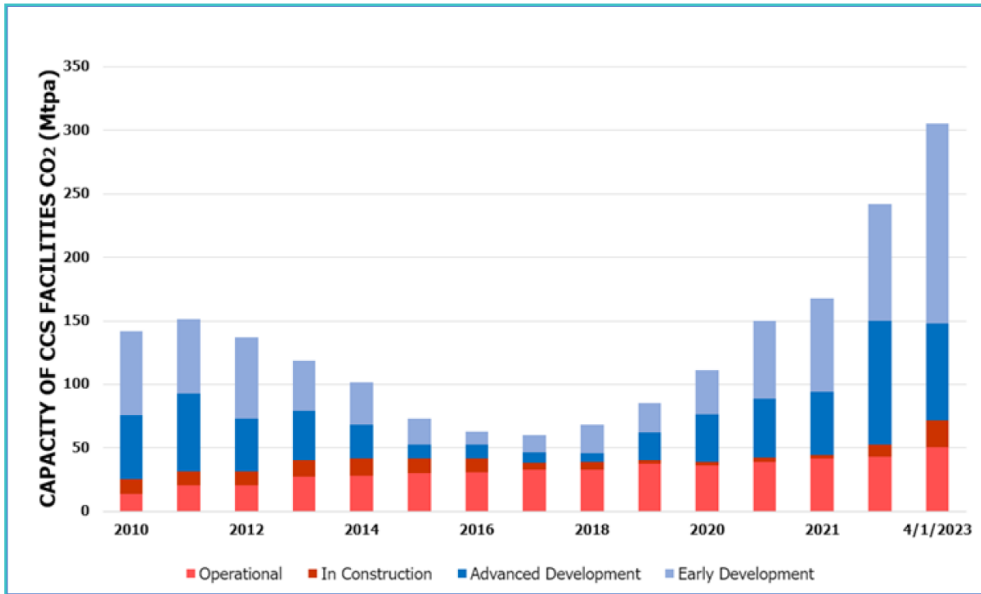
	2020	→	2050
Emission CO <sub>2</sub>	~40 Gt/y	→	~6 Gt/y
Absorption CO <sub>2</sub>	~0 Gt/y	→	~6 Gt/y

Need for increase development of Carbon reduction initiatives such as CCUS



# CARBON NEUTRALITY OBJECTIVES AND CARBON CAPTURE

## PIPELINE OF COMMERCIAL CCS FACILITIES BY CAPTURE CAPACITY



The capture capacity of facilities excludes:

- Transport and storage-only facilities
- Suspended operations
- Announced facilities

Data through 1 April 2023

Global CCS institute data

But more industry progress is needed to meet societal emissions goals

Global CCS industry capacity and project pipeline trends<sup>1</sup>, MTA



Only ~50 MTA of CCS capacity is operational today.

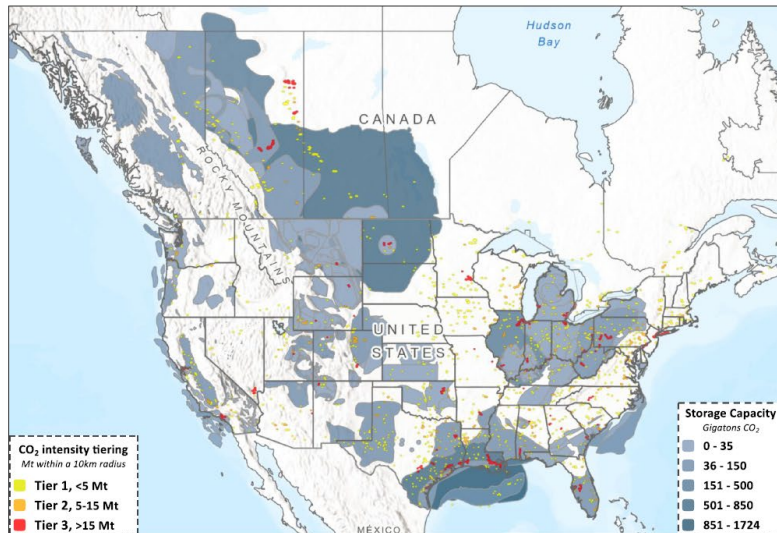
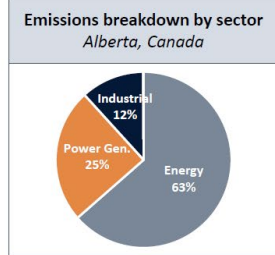
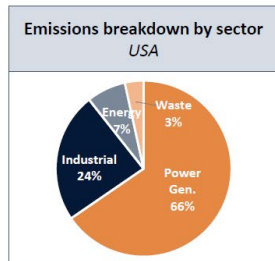
If all projects proceed, society will experience a 7x growth rate through 2030 but will be 3x short of the pathway needed to achieve IEA Net Zero Emissions 2050 Scenario.

We need to identify enablers for exponential acceleration.

# CCUS DEVELOPMENT CONSEQUENCES

→ Developing infrastructure for Capture, Transport and Storage in wide variety of regions in NA to connect large emitters (Oil/ Energy / Agro / Ethanol/ Steel/ Cement) to storage locations

CCUS is potentially as important to the 10 years as shale and LNG have been to the last 10 years



→ CO<sub>2</sub> transportation will cease to remain limited to « remote locations » and « short distance » direct injection for EOR to be closer to population.

→ Increase of CO<sub>2</sub> transportation associated risks (corrosion, leakage, asphyxiation)

# INCREASED RISKS OF ACCIDENTS SUCH AS THIS ONE

Feb 2020: GeoHazard → pipeline rupture, 350 m<sup>3</sup> liquid CO<sub>2</sub> released (385 mt), vaporized and plume heading to Satartia town : 49 people injured (intoxication) and 250 people evacuated



Figure 2: Vehicle is Parked on HWY 433 - The White is Ice Generated by the Release of CO<sub>2</sub> - The Blue Arrow Points North (Aerial Drone Photograph Courtesy of the Mississippi Emergency Management Agency)



Figure 6: Topographical Map Showing the Delhi Pipeline (Green) and Denbury's Buffer Zone (Red) on Either Side of the Pipeline and the Proximity to Satartia (Blue Star Indicates the Rupture Site)

Odorized CO<sub>2</sub> due to presence of H<sub>2</sub>S and other impurities (geological source of CO<sub>2</sub>) → enabled detection by public even though the cloud had moved away from pipeline

# Odorization technology

# INTRODUCTION TO ODORIZATION TECHNOLOGY AS APPLIED TO NATURAL GAS

→ **Why odorization:** to detect leakages for dangerous odorless gases/liquids to protect the population against leakage consequences

→ **Odorant characteristics** (cf ISO 13734):

- Strong odor at low concentration
- Distinctive « gassy » smell
- Non toxic
- Stable during storage and use with gas/liquid to be odorized

**Main odorants used for natural gas:**

- Europe/China/Maghreb/Singapore: Pure THT
- Americas/Middle East/Australia: TBM based blends
- EM not used but may be naturally present in nat gas

**Main odorants used for LPG: EM**

Table A.1 - List of chemical and physical properties of pure sulphur compounds

Sulfur compound	Formula	Molar mass g/mol	Boiling point °C	Freezing point °C	Density (at 20 °C) g/cm <sup>3</sup>
<b>Sulfides (thioether)</b>					
Dimethyl sulfide (DMS)	CH <sub>3</sub> SCH <sub>3</sub>	62,14	37,3	-98,3	0,848 3
Methyl ethyl sulfide (MES)	CH <sub>3</sub> SC <sub>2</sub> H <sub>5</sub>	76,16	66,7	-105,9	0,842 2
Diethyl sulfide (DES)	(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub> S	90,19	92,1	-103,9	0,836 2
Tetrahydrothiophene (THT)	C <sub>4</sub> H <sub>8</sub> S	88,17	121,0	-96,1	0,998 7
<b>Mercaptans (thiols)</b>					
Methylmercaptan (MM) <sup>a</sup> (methanethiol)	CH <sub>3</sub> SH	48,11	5,9	-123	0,866 5
Ethylmercaptan (EM) <sup>a</sup> (ethanethiol)	C <sub>2</sub> H <sub>5</sub> SH	62,14	35,1	-147,8	0,831 5 <sup>b</sup>
n-Propylmercaptan (NPM) (1-propanethiol)	C <sub>3</sub> H <sub>7</sub> SH	76,16	67 to 68	-113,3	0,841 1
iso-Propylmercaptan (IPM) (2-propanethiol)	(CH <sub>3</sub> ) <sub>2</sub> CHSH	76,16	52,6	-130,5	0,814 3
n-Butylmercaptan (NBM) (1-butanethiol)	C <sub>4</sub> H <sub>9</sub> SH	90,19	98,5	-115,7	0,841 6
sec.-Butylmercaptan (SBM) (2-butanethiol)	CH <sub>3</sub> CH(SH)C <sub>2</sub> H <sub>5</sub>	90,19	85	-165	0,829 5
iso-Butylmercaptan (IBM) (2-methylpropane-1-thiol)	(CH <sub>3</sub> ) <sub>2</sub> CHCH <sub>2</sub> SH	90,19	88,5	< -70	0,835 7
tert.-Butylmercaptan (TBM) (2-methylpropane-2-thiol)	(CH <sub>3</sub> ) <sub>3</sub> CSH	90,19	64,3	-0,5	0,794 3 <sup>b</sup>

Values taken from the *Handbook of Chemistry and Physics*, 87th ed., CRC Press, Boca Raton, Florida, USA.

# SAFETY APPROACH

→ Odorants are mainly used for flammable gases/liquids

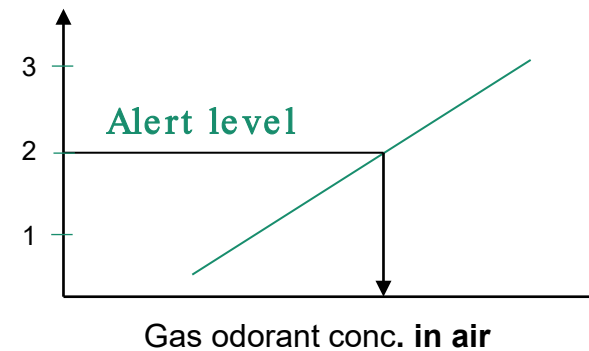
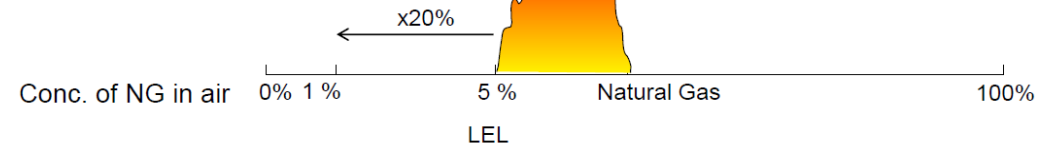
## SAFETY APPROACH

Alert at 20% of LEL

### Which odorant concentration?

- Intensity curves are established for each odorant blend (by trained people) according to a referenced method.
- Definition of a concentration allowing non trained people **to smell and to identify** the odor with a simple sniff (Sales scale level 2)

*Odor of Nat. Gas: Strong enough to be perceived by a normal nose at least when its concentration is equal at 20% of the lower explosive limit (LEL), which is 1% of gas in air*



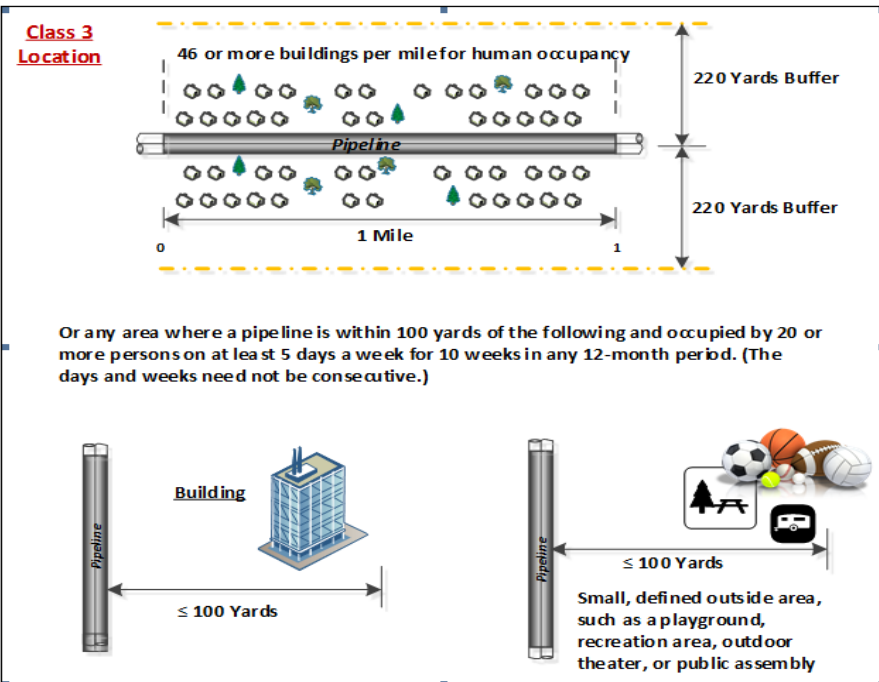


# US REGULATORY REQUIREMENT: IN POPULATED AREAS

→ Natural Gas Pipeline Safety Act issued 1968:

→ Enabled DOT- PHMSA to establish **49 CFR Part 192** in 1970, including the odorization requirements for distribution and transmission networks in **§192.625**

→ **American Gas Association (AGA) Odorization Manual**: describes the pipeline classification



→ **Odorization** being only required in **Distribution and Transmission class 3 & 4** pipelines

# ODORIZATION EFFICIENCY: CASE STUDY FOR FRANCE



→ Efficient odorization program enables about 28 000 calls\* for gas leakages per annum, enabling quick repairs and preventing as many gas accumulation situations.

- ~ 38 000 km (23 000 miles) high pressure natural gas transmission pipelines
- +
- > 200 000 km (124 000 miles) natural gas distribution network

> 150 000 miles all odorized using the same technology: TetraHydroThiophene

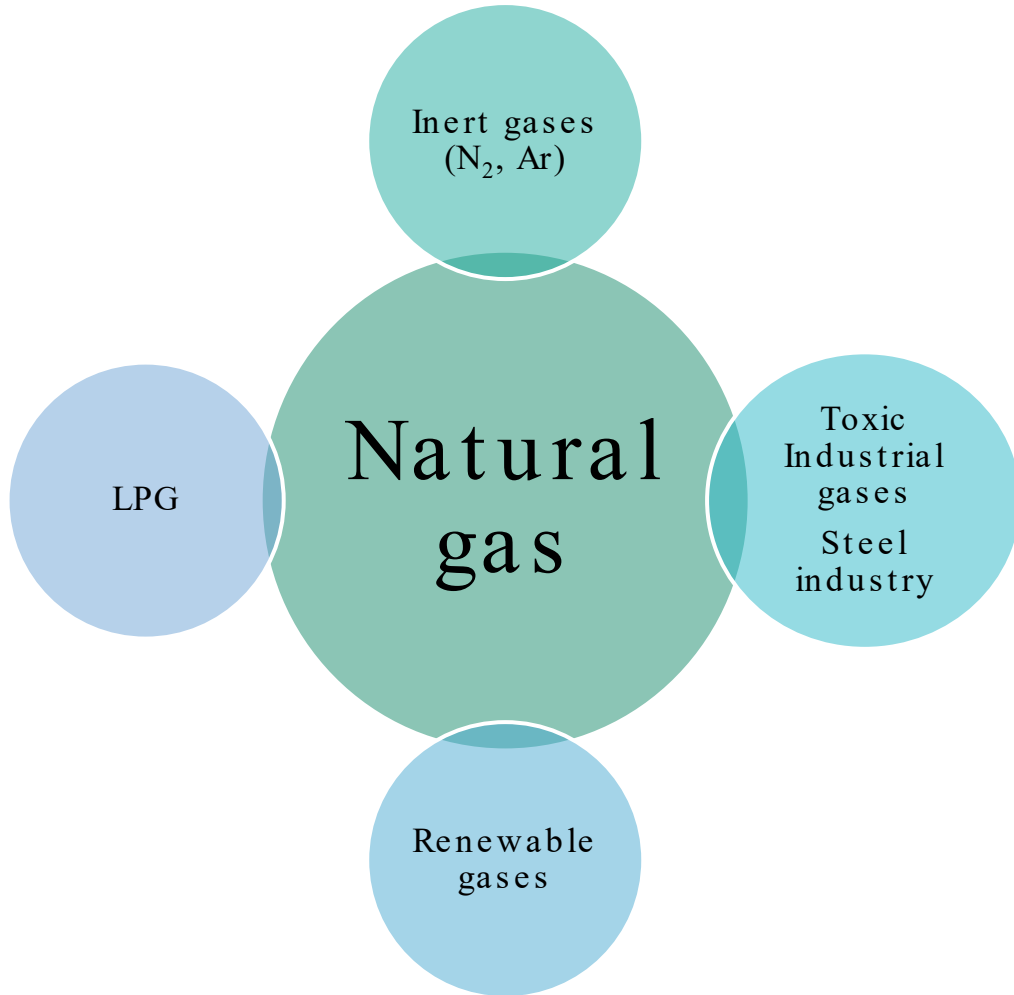


DIRECTION GÉNÉRALE  
DE LA SÉCURITÉ CIVILE  
ET DE LA GESTION DES CRISES

\* 2020 figures

Odorization  
Technology  
Transferrability

# ODORIZATION TECHNOLOGY TRANSFER : CASE STUDIES



Gas source	O <sub>2</sub>	CO	H <sub>2</sub>	CO <sub>2</sub>	N <sub>2</sub>
BFG	0,4%	20-36%	1,5- 8%	3,5- 13%	55- 61%
LDG	traces	65-80%	-	5- 7%	A little



# METHODOLOGY: EXEMPLE OF STEEL GASES

## RISK BASED EVALUATION

Definition of highest tolerated exposure considering gas composition and associated risks.

- *Toxicity risk of CO achieved before flammability and asphyxiation risks (few hundreds ppm CO)*
- *ALERT dosage determined to warn people before toxic exposure (STEL 300 ppm)*  
*Looking for traces → need for high intensity odorant*

## TECHNOLOGY PROPOSAL ACCORDING TO SCENARIO (evaluation of fading potential)

- Evaluation of Gas purity: presence of deleterious impurities, condensates, water, dusts.  
*Presence of dusts and water saturated*
- Evaluation of potential technical barriers: compatibility of end use with sulfur traces  
*No influence on steel surface, limited SOx emissions*
- Evaluation of process conditions: pressure, temperature, flowrate, pipe network material section and length  
*Short carbon steel networks at low pressure and high flowrates at ambient temperature*

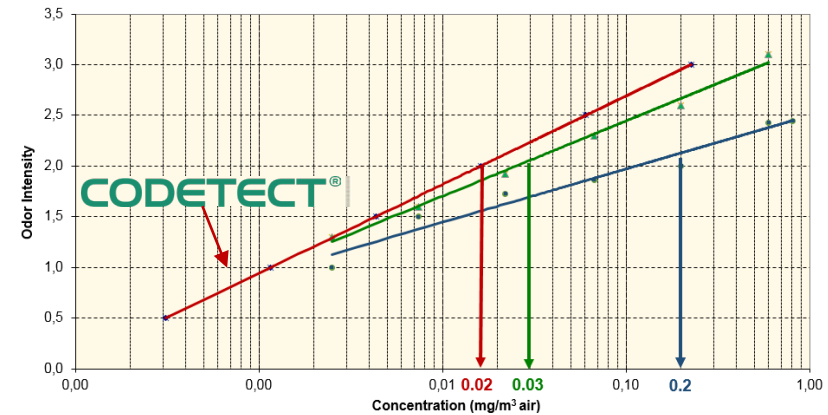
# METHODOLOGY: EXEMPLE OF STEEL GASES

## ARKEMA TECHNOLOGY PROPOSAL:

*Proprietary mercaptan / sulfide formulation with properties enabling low reactivity, high intensity, easy regasification after adsorption on surface or into condensates:*

**CODETECT®**

## DETERMINATION OF ODOR INTENSITY CURVE



## PILOTING

*Field evaluation: piloting in steel industry on BFG (20-30% CO) to assess odorization efficiency*

## IMPLEMENTATION

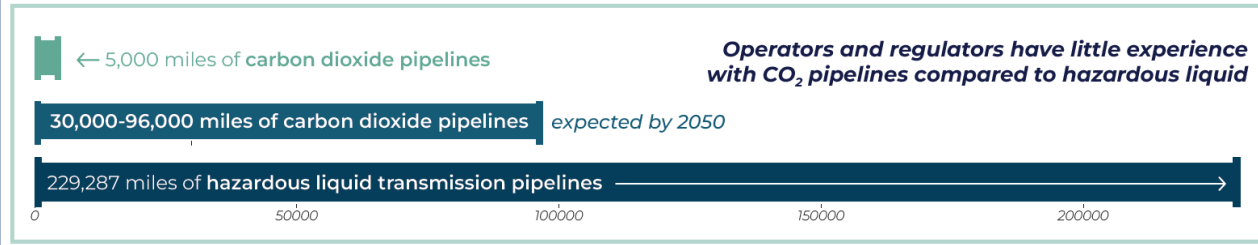
*Field implementation in China to improve safety around LDG networks (65-80% CO)*



# EVOLUTION OF CO<sub>2</sub> TRANSPORTATION NETWORK

- Existing CO<sub>2</sub> onshore transport for Enhanced Oil Recovery (EOR), but mainly in **remote areas** (eg Canada Quest) / over **short distances**
- With CCS more CO<sub>2</sub> will have to be transported onshore **through long distances** in **populated regions**

## CO<sub>2</sub> PIPELINE MILEAGE AND REGULATIONS



## CO<sub>2</sub> leakage risk in populated areas

→ Toxicity by Oxygen displacement

STEL = 30 000 ppm

Considering a Safety factor → 20 to 50% STEL

→ We could aim 6 000 - 15 000 ppm CO<sub>2</sub>

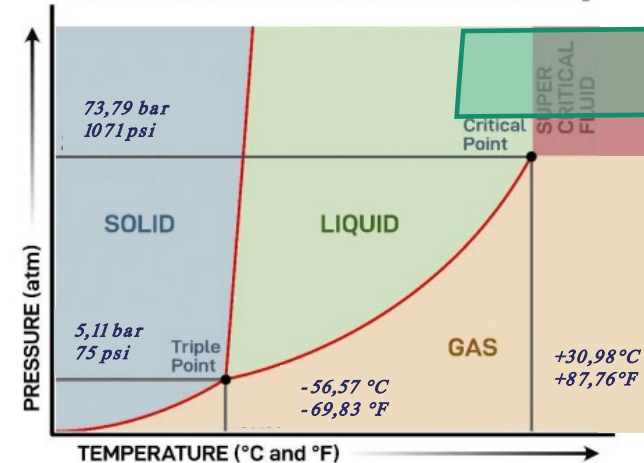
## Transport conditions:

- Gas phase
- Dense phase (supercritical or liquid)

*Supercritical* : pressure and temperature above critical point

- Transport phase determined by optimization of transport costs (pressure drop)

PHASE DIAGRAM OF CARBON DIOXIDE (CO<sub>2</sub>)



Max throughput:

Dense phase:

Pressure >74 bar (1071 psi)

# ODORIZATION TECHNOLOGY TRANSFER FOR CO<sub>2</sub> TRANSPORT

Pipeline conditions (Material, P, T) are acceptable with respect to odorant and current equipment used for injection in HP Transmission networks.

- Odorant **currently injected in up to 90 bars (1300psi)** pipelines in France. And **up to 250 bars (3600psi)** for CNG stations.
- Odorant should be **fully injectable down to -45°C (-50°F)** (NA Winter season)

They display **much lower freezing point** and ppm level water content resulting in **extremely low “cloud point”** (*temperature at which free water traces separate and crystallize*)

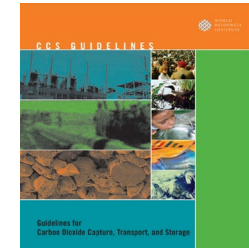
- Compatibility of the S-based odorant with usual pipeline material is already proven (**no corrosion issue**).

**refer to : ISO 13734 requirements**

**Odorants were already recommended to be considered in case of populated areas in 2008**

*(establishment of this report)*

[http://pdf.wri.org/ccs\\_guidelines.pdf](http://pdf.wri.org/ccs_guidelines.pdf)



**Satartia accident lessons:**

***Sulfur-based impurities in CO<sub>2</sub> stream can effectively odorize***

*Targeting specific odorant would bring benefits:*

***Standardized practice (Stability & Efficiency of odorant)***

***Improved Safety for population***



# R&D TO MEET CCS DEVELOPMENT NEEDS FOR IMPROVED SAFETY

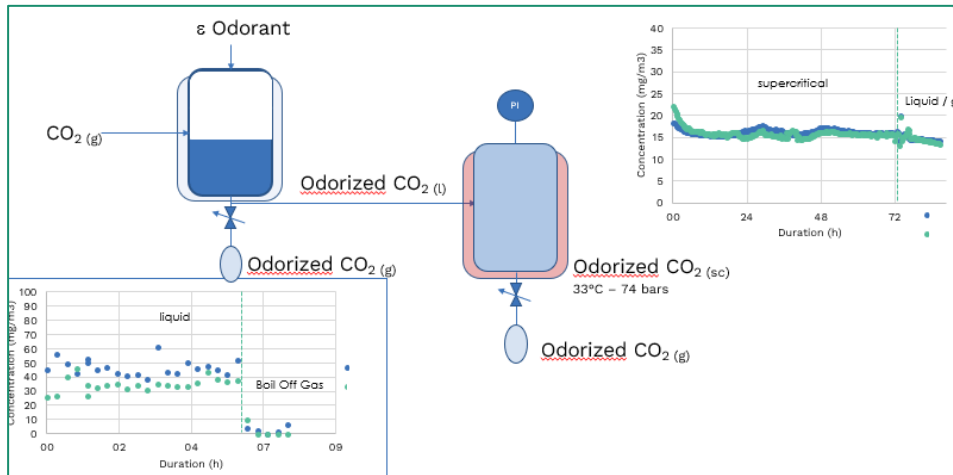
Technical feasibility evaluated under various conditions with pure CO<sub>2</sub>

Lab evaluation performed

*Arkema odorant R&D – Lacq – France*

Gaseous CO<sub>2</sub> odorization (**Pure**), liquefaction, and leak simulation from:

- Gas
- Liquid
- Supercritical state



## Impact of CO<sub>2</sub> impurity profile on technology selection

Intensity / odor character / stability

Example of some European projects spec

*Purity from > 91 % to > 96 %*

*Some impurities tracked down to ppm, other to % levels.*

*Several families of chemicals:*

- Hydrocarbons
- Oxygenates (ROH; RC(O)H; RC(O)OH)
- Amines
- S-derivatives

Evaluations to perform at R&D level or through Piloting.

	CO <sub>2</sub> specifications from different projects & operators					
	Northern Lights <sup>1</sup>	PORTHOS <sup>2</sup>	ARAMIS <sup>3</sup>		National Grid <sup>4</sup>	
			Shipping	Pipelines	Dense Phase	Gas Phase
CO <sub>2</sub>	Balance	≥ 95% mol	Balance	≥ 95% mol	≥ 96% mol	≥ 91% mol
H <sub>2</sub> O	≤ 30 ppm	≤ 70 ppm	≤ 30 ppm	≤ 70 ppm	≤ 50 ppm <sub>v</sub>	≤ 50 ppm <sub>v</sub>
O <sub>2</sub>	≤ 10 ppm	≤ 40 ppm	≤ 10 ppm	≤ 40 ppm	≤ 10 ppm <sub>v</sub>	≤ 10 ppm <sub>v</sub>
NO <sub>x</sub>	≤ 10 ppm	≤ 5 ppm	≤ 1.5 ppm	≤ 2.5 ppm	≤ 100 ppm <sub>v</sub>	≤ 100 ppm <sub>v</sub>
SO <sub>x</sub>	≤ 10 ppm		≤ 10 ppm		≤ 100 ppm <sub>v</sub>	≤ 100 ppm <sub>v</sub>
H <sub>2</sub> S	≤ 9 ppm		≤ 5 ppm		≤ 20 ppm <sub>v</sub>	≤ 80 ppm <sub>v</sub>
COS	-	≤ 20 ppm (sum) (of which H <sub>2</sub> S ≤ 5 ppm)	-	≤ 20 ppm (sum) (of which H <sub>2</sub> S ≤ 5 ppm)	-	-
(CH <sub>3</sub> ) <sub>2</sub> S	-	-	-	-	-	-
Dimethyl sulfide	-	-	-	-	-	-
H <sub>2</sub>	≤ 50 ppm	≤ 0.75% mol	≤ 500 ppm	≤ 0.75% mol	≤ 2% mol	≤ 2% mol
N <sub>2</sub>	‡	≤ 2.4% mol	-	≤ 2.4% mol	*	-
Ar	‡	≤ 0.4%	-	≤ 0.4% mol	*	-
CH <sub>4</sub>	‡	≤ 1%	-	≤ 1%	*	-
CO	≤ 100 ppm	≤ 750 ppm	0.12% mol	≤ 750 ppm	≤ 2000 ppm <sub>v</sub>	≤ 2000 ppm <sub>v</sub>
O <sub>2</sub> +N <sub>2</sub> +H <sub>2</sub> +Ar+CH <sub>4</sub> +CO		≤ 4%	< 2000 ppm	< 40000 ppm	-	-
Amine	≤ 10 ppm	≤ 1 ppm	≤ 10 ppm	≤ 1 ppm	‡	‡
NH <sub>3</sub>	≤ 10 ppm	≤ 3 ppm	≤ 10 ppm	≤ 3 ppm	‡	‡
HCN	-	≤ 2 ppm	-	≤ 2 ppm	‡	‡
Formaldehyde	≤ 20 ppm	-	≤ 20 ppm	-	-	-
Acetaldehyde	≤ 20 ppm	-	≤ 20 ppm	-	-	-
Total aldehydes	-	≤ 10 ppm	-	≤ 10 ppm	-	-
C <sub>2</sub> +hydrocarbons	-	≤ 0.12% mol	-	≤ 0.12% mol	-	-
Aromatics	-	≤ 0.1 ppm	-	≤ 0.1 ppm	-	-
C <sub>6</sub> H <sub>6</sub>	-	n/a	-	-	-	-
Total VOC	-	≤ 10 ppm	≤ 10 ppm	≤ 10 ppm	-	-
Total glycol compounds	-	Follow dew point spec.	-	Follow dew point spec.	-	-
Total carboxylic acid and amide compounds	-	≤ 1 ppm	-	≤ 1 ppm	-	-
Tot P contained compounds	-	≤ 1 ppm	-	≤ 1 ppm	-	-
Ethanol	-	≤ 20 ppm	≤ 20 ppm	≤ 20 ppm	-	-
Methanol	-	≤ 620 ppm	≤ 40 ppm	≤ 620 ppm	-	-
Mercury, Hg	≤ 0.03 ppm	-	≤ 0.03 ppm	-	‡	‡

QUESTIONS ?  
PLEASE CONTACT US

## REFERENCES

### ODORIZATION STANDARDS AND REFERENCE MANUALS

- AGA Odorization Manual
- ISO 13734
- ISO 16922

### ARKEMA GAS ODORANT

### PRODUCT & SERVICE OFFER



### LITERATURE:

*2013 QUEST Project documentation: odorant injection study*

*2014 review on CO<sub>2</sub> transportation*  
GHGT-12

CO<sub>2</sub> Pipeline infrastructure – lessons learnt  
10.1016/j.egypro.2014.11.271

*2015: Odourisation of CO<sub>2</sub> pipelines in the UK: Historical and current  
impacts of smell during gas transport*  
10.1016/j.ijggc.2015.04.010

*2022: Failure Investigation Report – Denbury Gulf Coast Pipelines LLC*

CCS Guidelines(2008):  
[http://pdf.wri.org/ccs\\_guidelines.pdf](http://pdf.wri.org/ccs_guidelines.pdf)

Technical and Economic Characteristics of a CO<sub>2</sub> Transmission Pipeline  
Infrastructure (European commission-2011)

*2023: Pipeline Safety Trust*  
CO<sub>2</sub> Pipeline Safety : summary for policy makers (May 2023)